## B.Sc. 5th Semester (Honours) Examination, 2023 (CBCS)

**Subject: Chemistry** 

Course: DSE-2

## (Instrumental methods of Chemical Analysis)

Time: 2 Hours

Full Marks: 40

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable.

Answer any five questions from the following:

 $2 \times 5 = 10$ 

- (a) What are the light sources used in UV-vis spectrophotometer?
- (b) Why is quartz cuvette used to measure spectra in UV-vis and fluorescence spectrophotometer?
- (c) Why tetramethyl silane (TMS) is used as an internal standard in NMR spectroscopy?
- (d) What are the light sources used in spectrofluorometer?
- (e) Define 'ionization chamber' in mass spectrometry.
- (f) Which of <sup>13</sup>C and <sup>12</sup>C is NMR active? State reason.
- (g) Define signal to noise ratio (S/N) in spectroscopic technique.
- (h) What type of cell in used to measure IR spectra of hygroscopic liquid/aqueous samples?
- 2. Answer any two questions from the following:

 $5 \times 2 = 10$ 

- (a) Draw a block diagram of double beam UV spectrophotometer. Explain the monochromator device. 4+1
- (b) Write down the differences between liquid and gas chromatography? What is the difference between (+) and (-) mode of ionisation in mass chamber? 3+2
- (c) Draw the various fragmented patterns of *n*-butyl alcohol and *sec*-butyl alcohol.

3+2

- (d) What are the functions of using external standard in NMR? What are the advantages of Fourier Transform in Infrared Spectroscopy?
- 3. Answer any two questions from the following:

 $10 \times 2 = 20$ 

- (a) Write down the block diagram of instrumentation of mass spectrometer. Define the term 'chemical shift'. Which one of 'mass spectrometer' and 'mass spectroscopy' is correct? Defend your answer. 4+3+1+2
- (b) Describe the principle of Atomic Absorption Spectroscopy (AAS)? Which gases are used in AAS? What are the differences between single and double beam spectrophotometer? 5+2+3

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- (c) Name the fuels used in flame photometry. Discuss briefly about the columns used in liquid chromatographic analysis. Define  $R_f$  value in chromatographic analysis. What is the sampling procedure in IR spectrophotometer? 2+5+2+1
- (d) 'Increase in polarity of the solvent shifts  $n \to \pi^*$  and  $n \to \sigma^*$  bands to shorter wavelength'—comment on the statement. Briefly describe the spin-spin splitting in NMR spectroscopy. 'FT-IR is plotted against wavenumber instead of wavelength'—explain. 4+4+2